

Figure 10. Sub-slab ventilation using an opening cut through the slab and into the sub-slab gravel. This is simpler than the case of a sump in a sump opening but more than one opening may have to be made to ensure complete sub-slab ventilation without sub-slab drainpipes.

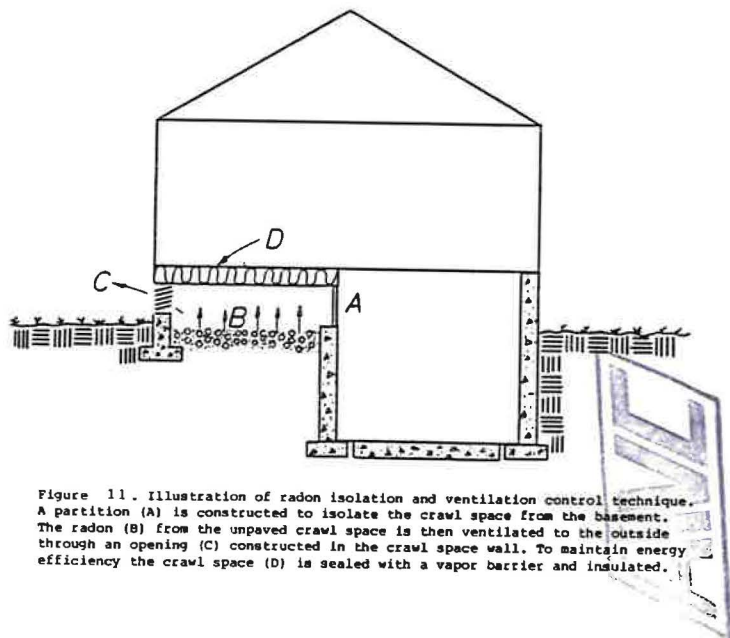


Figure 11. Illustration of radon isolation and ventilation control technique. A partition (A) is constructed to isolate the crawl space from the basement. The radon (B) from the unpaved crawl space is then ventilated to the outside through an opening (C) constructed in the crawl space wall. To maintain energy efficiency the crawl space (D) is sealed with a vapor barrier and insulated.

# PRELIMINARY EVALUATION OF FORMALDEHYDE MITIGATION STUDIES IN UNOCCUPIED RESEARCH HOMES



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The effectiveness of retrofit formaldehyde ( $\text{CH}_2\text{O}$ ) mitigation measures for energy efficient homes is being investigated in unoccupied research houses constructed according to East Tennessee building codes. Formaldehyde emissions from carpet-covered, particleboard underlayment throughout these houses have frequently caused indoor  $\text{CH}_2\text{O}$  concentrations to exceed 0.1 ppm comfort guidelines, particularly during warm and humid seasons. The effectiveness of carpet and cushion, vinyl linoleum, and polyethylene vapor barriers over the underlayment have been compared with increased ventilation for reduction of indoor  $\text{CH}_2\text{O}$  concentrations under controlled 23°C and 50% relative humidity (RH) conditions. Approximate 2 to 2.5 fold reductions in  $\text{CH}_2\text{O}$  concentrations were achieved with the linoleum and polyethylene barriers. Simple steady-state models predict that sevenfold increases in air exchange rate would be required for comparable improvements in  $\text{CH}_2\text{O}$  levels. In contrast, common nylon carpet and urethane foam cushion flooring materials were ineffective in reducing  $\text{CH}_2\text{O}$  levels.

Surface emission measurements from a selected floor area were performed with each floor barrier to compare reductions in CH<sub>2</sub>O emissions specifically from the underlayment with reductions in indoor CH<sub>2</sub>O concentrations, which are complicated by additional CH<sub>2</sub>O sources and sinks. Approximate 10 to 30 fold decreases in underlayment emissions were measured with linoleum and polyethylene barriers in comparison to minimal changes with carpet and cushion materials. The comparatively small 2 to 2.5 fold decrease in indoor CH<sub>2</sub>O concentrations (with low permeability floor barriers) is presumably caused by the presence of secondary sources and their increased CH<sub>2</sub>O emission rates at reduced CH<sub>2</sub>O concentrations.

## Introduction

Formaldehyde (CH<sub>2</sub>O) is an important indoor air pollutant that is emitted by a variety of consumer and construction products including pressed-wood products, insulation materials, textiles, and combustion sources (1). The strongest contributors to indoor CH<sub>2</sub>O are typically urea-formaldehyde foam insulation and pressed-wood products, such as particleboard underlayment, hardwood plywood paneling, and medium density fiberboard, that are fabricated with urea-formaldehyde resins (2). These products typically have the strongest CH<sub>2</sub>O emission rates, the largest product loading [source area (m<sup>2</sup>)/compartment volume (m<sup>3</sup>)], and are frequently covered with CH<sub>2</sub>O-permeable physical barriers.

Control strategies for CH<sub>2</sub>O in indoor environments are receiving increased attention, particularly in industrial research efforts to reduce CH<sub>2</sub>O emissions from pressed-wood products through improvements in resin and/or board fabrication technologies (3). The effectiveness of retrofit mitigation measures for individual and multiple CH<sub>2</sub>O emission sources in indoor environments have recently been surveyed and modeled (2). Product aging for greater than five to ten years and complete removal of all major emission sources were found to be effective but impractical control measures. Increased air exchange rates reduced indoor CH<sub>2</sub>O concentrations resulting from all indoor emission sources. However, the modeled increase in CH<sub>2</sub>O emission rates from CH<sub>2</sub>O resin-containing products at reduced CH<sub>2</sub>O concentrations made large improvements in indoor levels energy-costly, particularly during heating and cooling seasons. Tenfold reductions in CH<sub>2</sub>O emissions from individual sources have been achieved in laboratory experiments with low permeability barriers such as vinyl linoleum (4). Threefold to tenfold reductions in CH<sub>2</sub>O concentrations inside manufactured housing have been experimentally achieved over 0.5 to 1 year periods using ammonia fumigation techniques (2).

Indoor air quality studies, aimed at the control of CH<sub>2</sub>O levels in energy efficient homes, have been conducted in three-bedroom, unfurnished research houses constructed in 1981 according to East Tennessee building codes. Interhouse comparison measurements of CH<sub>2</sub>O, radon, and air infiltration rates in 1982-4 indicated that the houses were well matched and had CH<sub>2</sub>O concentrations in excess of the American Society of Heating and Air Conditioning Engineers (ASHRAE) guideline of 0.1 ppm (5). The impact of variation in indoor temperature and relative humidity (RH) on CH<sub>2</sub>O concentrations inside two research houses were measured and modeled in

1984 and compared against the results of prior laboratory studies of particleboard underlayment, the primary emitter in the test houses. The results of the study indicated fourfold variation in CH<sub>2</sub>O concentration from 0.07 to 0.27 ppm with potential seasonal conditions of 20°C, 30% RH and 29°C, 80% RH, respectively, consistent with predictions using selected laboratory models (6). The control of indoor CH<sub>2</sub>O concentrations below 0.1 ppm by maintaining a cool and dry indoor climate was possible in winter but energy-costly in warmer seasons.

A second study has been performed in a single research house to compare the effectiveness of increased air exchange rates and various retrofit barriers over the particleboard underlayment in reducing indoor CH<sub>2</sub>O levels. Measurements of CH<sub>2</sub>O concentration were performed as a function of floor barrier and ventilation rate inside the conditioned envelope of the home and inside a 0.2 m<sup>3</sup> chamber that was used as a surface emission monitor over the floor. Thus, the reduction in CH<sub>2</sub>O emission rates from the flooring-covered underlayment can be compared against the overall impact on indoor CH<sub>2</sub>O concentrations, which is complicated by additional CH<sub>2</sub>O emission sources and sinks.

## Experimental Design

The CH<sub>2</sub>O mitigation measures study was performed using whole house retrofits of the carpet and cushion flooring, which covered particleboard underlayment over about 85% of the accessible floor area (see Figure 1).

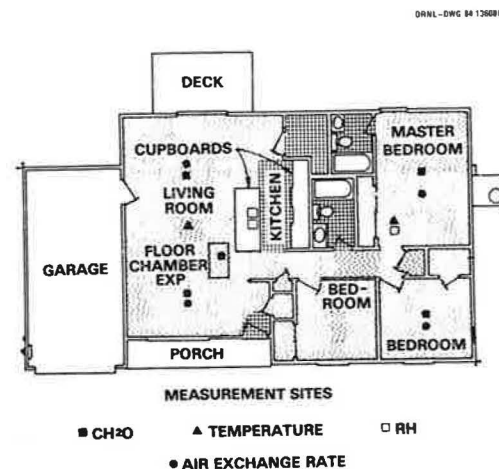


Figure 1: Physical layout and measurement sites inside the research house.

The remaining visible floor area, found primarily in the bathrooms and kitchen, was covered with vinyl linoleum and left unaltered during the mitigation measures study. The order of testing for the different floor-barrier systems was (1) the original nylon carpet and 1.3 cm neoprene sponge rubber cushion, (2) bare underlayment, (3) vinyl linoleum, (4) 6 mil polyethylene sheet, and (5) repeat measurements with the original carpet and replacement cushion. Standard nailstrip construction was used for the installation of carpet following the stapling of the carpet cushion to the underlayment. Linoleum and polyethylene sheet were installed with staples, double sided tape at joints, and a narrow strip of glue under the edges near walls. Such practice provided a continuous seal to the underlayment and easy removal without significant alteration to the surface of the underlayment.

Separate experimental designs were used to evaluate the impact of different flooring barriers on (1) the CH<sub>2</sub>O concentrations inside the conditioned envelope of the research house, and (2) the CH<sub>2</sub>O emission rates specifically from the particleboard underlayment. In the house experiments, the impact of floor barriers on the CH<sub>2</sub>O emission rate from the underlayment were examined indirectly through variations in indoor CH<sub>2</sub>O concentrations. The surface emission measurements of the underlayment and barrier combinations examined the impact of specific barriers on the CH<sub>2</sub>O emission rate from the underlayment in the absence of secondary sources and CH<sub>2</sub>O sinks that are found inside the house.

Both experimental designs involved controlled temperature, RH, air exchange rates, and atmospheric mixing. For the whole house experiment, temperature and RH control were maintained at typically 23 ± 1°C and 50 ± 2% RH with the central HVAC system and several humidifiers/dehumidifiers, respectively. Intercompartment mixing was enhanced using several 0.5 m fans and by keeping interior doors open. The air exchange rate of the house was adjusted to levels of approximately 0.2 h<sup>-1</sup> (i.e., residual level), 0.5 h<sup>-1</sup> and 1.0 h<sup>-1</sup> using the HVAC circulation fan, bathroom exhaust fans, and an adjustable fan exhausting into the attic. For the surface emission experiment, the temperature inside the chamber was controlled by the house temperature. The RH and air exchange rates inside the chamber were controlled using a metered air flow that was conditioned with water vapor to approximately 50% RH (7). The air exchange rates through the chamber (i.e., 0.6, 1.5 and 3.0 h<sup>-1</sup>) were chosen to achieve CH<sub>2</sub>O concentrations similar to those inside the house with the original carpet and cushion floor barrier. This procedure provided a similar driving force (i.e., CH<sub>2</sub>O concentration gradient) for CH<sub>2</sub>O diffusion from the underlayment into the house and chamber. Mixing of the chamber atmosphere was enhanced with an 8 cm fan and the pumped air supply.

The sequential investigation of the test floor barriers for both the whole house and surface emission experimental designs involved systematic conditioning and measurement of the interior of the house and the floor-chamber micro-environment. After a floor-barrier was installed, the house and chamber were conditioned at approximately 23°C and 50% RH for typically one to two weeks prior to CH<sub>2</sub>O measurements. In addition, controlled air exchange rates were established inside the house and chamber for three to six days prior to measurements under each ventilation condition. This dual

conditioning protocol was used to establish quasi-steady-state CH<sub>2</sub>O concentrations inside the floor chamber and in the conditioned envelope of the house, where the time-dependent CH<sub>2</sub>O concentration was complicated by numerous CH<sub>2</sub>O sources and sinks. Measurements of the temperature were performed at two floor-level locations at opposite ends of the house (see Figure 1). Measurements of air exchange rate and CH<sub>2</sub>O concentrations were performed at opposite ends of the house approximately one meter above the floor. Split sampling lines were used at each end of the house to sample simultaneously from two locations. Relative humidity data were recorded from a single location. However, hygrometers at multiple locations throughout the house indicated typically less than 1 to 2% RH variation. Only CH<sub>2</sub>O concentration measurements were performed for the floor chamber.

A one to four day measurement period was used for most floor barrier systems at each of three ventilation conditions for the floor chamber (i.e., 0.6, 1.5, and 3.0 h<sup>-1</sup>) and house (i.e., approximately 0.2, 0.5, and 1.0 h<sup>-1</sup>). Near-continuous temperature and RH data were taken in the house using thermocouples and a General Eastern Model 400E electrochemical sensor, respectively. Near-continuous air exchange rate measurements of the house were performed using a Miran Model 1A infrared spectrometer to monitor the time-dependent decline in concentration of injected freon gas (8). Formaldehyde concentration measurements were limited to a 15 minute sampling period per hour in each of the two house locations and the floor chamber. A modified CEA Instrument (9) and a gas switching manifold with teflon-lined valves and teflon tubing were used to sample and measure CH<sub>2</sub>O from all three measurement locations. Instruments were generally calibrated at the beginning of measurements for each floor-barrier system.

## Results and Discussion

The experimental results for the house and floor-chamber measurements of the CH<sub>2</sub>O mitigation study in the research house are reported in Table 1. Average parameter values are listed for each combination of floor barrier and ventilation condition. Temperature, air exchange rate, and CH<sub>2</sub>O concentration averages for multiple locations inside the research house are reported. The site to site variation of mean temperature, air exchange rate, and CH<sub>2</sub>O concentration were less than 0.7°C, 0.02 h<sup>-1</sup> and 0.004 ppm, respectively.

To distinguish the impact of the floor barriers on the measured CH<sub>2</sub>O concentrations inside the research house from that of the indoor climate and other time-dependent factors, two correction equations are considered. The whole-house CH<sub>2</sub>O concentration data are first normalized to standard test conditions of 23°C and 50% RH using a CH<sub>2</sub>O concentration model developed in previous studies of the temperature and RH dependence of CH<sub>2</sub>O concentrations inside the research houses (6).

$$[\text{CH}_2\text{O}]_{23^\circ\text{C}, 50\% \text{RH}} = \frac{[\text{CH}_2\text{O}]_{T, \text{RH}}}{e^{-6700 \cdot (1/T - 1/296)} \cdot (\text{RH}/50)^{0.86}} \quad (1)$$



Table 1. Average results from house and floor-chamber measurements as a function of floor-barrier.

A. Measurements of the Conditioned Envelope						
Floor Barrier	Elapsed Time (days)	Temperature (°C)	RH (%)	Air Exchange Rate (h <sup>-1</sup> )	[CH <sub>2</sub> O] (ppm)	Corrected <sup>a</sup> [CH <sub>2</sub> O] (ppm)
Carpet, Cushion I	0	24.6 ± 0.4	52.5 ± 0.8	0.20 ± 0.05	0.108 ± 0.005	0.091
	6	23.6 ± 0.4	51.0 ± 1.7	0.50 ± 0.09	0.089 ± 0.013	0.085
	11	23.4 ± 0.2	52.9 ± 1.4	0.94 ± 0.17	0.062 ± 0.007	0.059
Bare Underlayment	32	23.5 ± 0.4	49.6 ± 1.7	0.25 ± 0.05	0.092 ± 0.008	0.099
	38	22.6 ± 0.9	49.0 ± 1.0	0.46 ± 0.07	0.070 ± 0.013	0.082
	46	22.3 ± 1.0	47.6 ± 2.0	0.98 ± 0.11	0.043 ± 0.012	0.053
Vinyl Linoleum	64	22.8 ± 0.5	51.4 ± 0.9	0.24 ± 0.06	0.036 ± 0.004	0.045
	58	23.1 ± 0.7	53.2 ± 0.5	0.48 ± 0.06	0.029 ± 0.006	0.033
	54	23.2 ± 0.6	52.3 ± 0.7	0.87 ± 0.18	0.022 ± 0.003	0.024
6 mil Polyethylene	74	24.1 ± 0.8	50.5 ± 2.1	0.21 ± 0.04	0.044 ± 0.006	0.053
	69	22.9 ± 0.5	51.3 ± 0.9	0.80 ± 0.24	0.024 ± 0.002	0.030
Carpet, Cushion II	84	23.1 ± 0.2	50.5 ± 0.4	0.20 ± 0.04	0.071 ± 0.004	0.097
	90	23.4 ± 0.4	49.9 ± 0.7	0.54 ± 0.05	0.055 ± 0.007	0.072
	96	23.4 ± 0.5	50.3 ± 0.7	1.18 ± 0.34	0.042 ± 0.006	0.055

B. Floor Surface Emission Measurements

Floor Barrier	Air Exchange Rate (h <sup>-1</sup> )	[CH <sub>2</sub> O] (ppm)	CH <sub>2</sub> O Emission Rate (mg/m <sup>2</sup> h)
Carpet and Cushion I	0.60 ± 0.03	0.127 ± 0.008	0.060 ± 0.005
	1.50 ± 0.08	0.078 ± 0.006	0.093 ± 0.009
	3.00 ± 0.15	0.060 ± 0.004	0.143 ± 0.012
Bare Underlayment	0.60 ± 0.03	0.124 ± 0.018	0.059 ± 0.009
	1.50 ± 0.08	0.083 ± 0.012	0.098 ± 0.015
	3.00 ± 0.15	---	---
Vinyl Linoleum	0.60 ± 0.03	0.014 ± 0.002	0.006 ± 0.001
	1.50 ± 0.08	0.008 ± 0.002	0.010 ± 0.003
	3.00 ± 0.15	0.003 ± 0.003	0.008 ± 0.007
6 Mil Polyethylene	0.60 ± 0.45	0.025 ± 0.002	0.012 ± 0.001
	1.50 ± 0.08	0.007 ± 0.003	0.009 ± 0.003
	3.00 ± 0.15	0.002 ± 0.001	0.005 ± 0.003
Carpet and Cushion II	0.60 ± 0.03	0.120 ± 0.005	0.057 ± 0.004
	1.50 ± 0.08	0.092 ± 0.006	0.109 ± 0.009
	3.00 ± 0.15	0.059 ± 0.008	0.139 ± 0.020

<sup>a</sup>See text for description of temperature, RH and time dependent corrections applied.

To estimate the impact of time-dependent factors, such as product aging or outdoor environmental effects, on the CH<sub>2</sub>O concentrations inside the research house during the course of the permeation barriers study, repeat measurements with a carpet and cushion barrier were performed at the end of the study. The results from the floor-chamber experiments show no systematic increase or decrease in measured CH<sub>2</sub>O concentrations between the beginning and end of the study. This indicates there was little change in the CH<sub>2</sub>O emission strength of the underlayment due to product aging. In contrast, the results for the house experiments show consistent 30-40% decreases in CH<sub>2</sub>O concentrations between the initial and final measurements with the carpet/cushion barrier at air exchange rates of approximately 0.2, 0.5, and 1.0 h<sup>-1</sup>. The cause for this decline in CH<sub>2</sub>O levels is uncertain but may be related to decreasing outdoor temperatures during the progression of the study from 09/84 to 12/84. The cooling of the crawl space may have caused the bottom side of the underlayment to be exposed to progressively lower temperatures, particularly near the perimeter of the home. Such lower temperature exposures could reduce the average CH<sub>2</sub>O emission strength of the underlayment and, therefore, indoor CH<sub>2</sub>O concentrations while relatively constant CH<sub>2</sub>O levels could be measured in the floor chamber experiment near the center of the home where temperature sensors were located and best temperature control was achieved. As a result, the CH<sub>2</sub>O concentration data for the whole house experiments were normalized to those values determined at the beginning of the study assuming a linear decline in CH<sub>2</sub>O concentration as a function of elapsed time between the initial and final measurements with the carpet/cushion barrier (see Appendix). The time-dependent correction factors for the results of the bare underlayment and linoleum and polyethylene barriers are typically less than 1.25.

The CH<sub>2</sub>O concentration data for the house experiments that are corrected for variation in indoor temperature, RH, and other time dependent factors are listed in Table 1 and illustrated in Figure 2. Little variation is observed between the bare underlayment and underlayment covered with nylon carpet and neoprene sponge rubber cushion at air exchange rates of approximately 0.2, 0.5, and 1.0 h<sup>-1</sup>. In contrast, approximate 2 to 2.5 fold reductions in CH<sub>2</sub>O concentrations are indicated with both the vinyl linoleum and 6 mil polyethylene sheet barriers over an approximate 0.2 to 1.0 h<sup>-1</sup> range of air exchange rate. These findings are generically confirmed in the results of the floor-chamber experiments. Minimal differences in measured CH<sub>2</sub>O emission rates are observed between the bare underlayment and carpet/cushion covered underlayment. In contrast, 10 to 20 fold and 5 to 30 fold reductions in CH<sub>2</sub>O emission rates are observed with vinyl linoleum and 6 mil polyethylene sheet.

The large reductions in CH<sub>2</sub>O emission rates from the linoleum- and polyethylene-covered underlayment measured in the floor-chamber experiments in comparison to the moderate decreases in indoor CH<sub>2</sub>O concentrations is evidence for (1) the existence of CH<sub>2</sub>O emitters other than the underlayment inside the home, and (2) enhanced CH<sub>2</sub>O emissions from these secondary emitters under reduced CH<sub>2</sub>O concentrations. The potential contribution of CH<sub>2</sub>O due to convective transport via infiltration from the underside of the underlayment is unknown. A survey of potential secondary CH<sub>2</sub>O emitters was performed by selective removal of product samples for laboratory

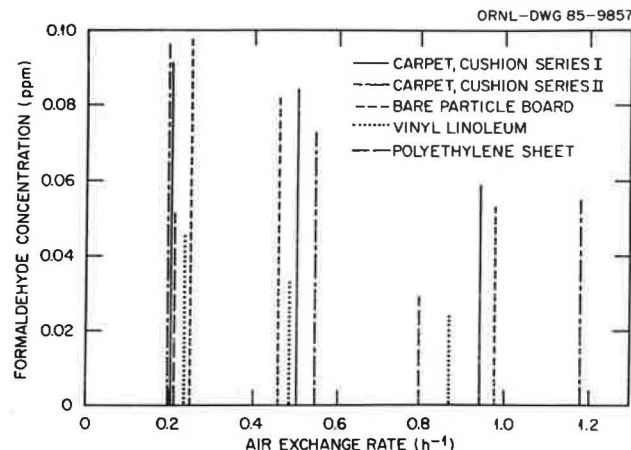


Figure 2: Average indoor formaldehyde concentrations as a function of floor barrier and air exchange rate. Data corrected to 23 C, 50% RH and an elapsed time of zero for the barriers study.

conditioning and measurement using a Formaldehyde Surface Emission Monitor (10). Only the particleboard and hardwood paneling products incorporated in counter tops and cabinet shelves had  $\text{CH}_2\text{O}$  emission rates (ie., 0.04-0.11  $\text{mg}/\text{m}^2\text{h}$ ) comparable to those of the underlayment. The carpet, carpet cushion, and most of the decorative exterior of the cabinets were insignificant emitters (ie., typically  $<0.01 \text{ mg}/\text{m}^2\text{h}$ ). The remaining  $\text{CH}_2\text{O}$  resin containing products are fibrous glass insulation and softwood plywood subflooring. These products are fabricated with phenol formaldehyde resins and are anticipated to be minimal contributors to indoor  $\text{CH}_2\text{O}$  concentrations (2). An additional source of  $\text{CH}_2\text{O}$  during the permeation barriers study, was the uncovered areas of underlayment near walls where the floor barriers could not be effectively applied. The combined area of particleboard near walls, in cabinet shelves and under counter tops is estimated to be about  $11 \text{ m}^2$ , approximately 15% of the area of the carpet/cushion covered underlayment. Although the contribution of these secondary emitters is anticipated to be small in the presence of the carpet/cushion covered underlayment, their contribution is enhanced at the reduced concentrations resulting from the linoleum or polyethylene barriers over the underlayment. At reduced  $\text{CH}_2\text{O}$  concentrations, the driving force for  $\text{CH}_2\text{O}$  transport from the particleboard to the surrounding atmosphere is enhanced. The negative  $\text{CH}_2\text{O}$  concentration dependence of the  $\text{CH}_2\text{O}$  emission rate (ER) from pressed-wood products has been modelled according to principles of molecular diffusion (11),

$$\text{ER} (\text{mg}/\text{m}^2\text{h}) = -A(\text{m}/\text{h}) \cdot [\text{CH}_2\text{O}] (\text{mg}/\text{m}^3) + B (\text{mg}/\text{m}^2\text{h}) \quad (2)$$

where A and B are constants. As a result of this diffusion effect, the impact of large reductions in  $\text{CH}_2\text{O}$  emission rate from a single source (through source removal or the use of effective permeation barriers) is lessened in the presence of other  $\text{CH}_2\text{O}$  emitters. A detailed treatment of the research houses using Equation 2 could not be performed. The requisite  $\text{CH}_2\text{O}$  emission rate data as a function of  $\text{CH}_2\text{O}$  concentration for emitters other than the underlayment was unavailable to determine the A and B constants.

The results of the permeation barriers study provide the necessary data to compare the effectiveness of various permeation barriers versus increased air exchange rates for reduction of  $\text{CH}_2\text{O}$  concentrations in the research houses. The requisite air exchange rate to achieve fixed reductions in  $\text{CH}_2\text{O}$  can be estimated using the following single compartment model (11,12),

$$[\text{CH}_2\text{O}]^{-1} (\text{m}^3/\text{mg}) = [1.8 \pm 0.1 (\text{m}^2\text{h}/\text{mg}) \cdot N (\text{h}^{-1}) / 0.29 \pm 0.02 (\text{m}^{-1})] + 7.5 \pm 0.2 (\text{m}^2/\text{mg}) \quad (3)$$

where N is the air exchange rate ( $\text{h}^{-1}$ ) and 0.29 is the loading ( $\text{m}^2/\text{m}^3$ ) of carpet/cushion covered underlayment, which is assumed to be the only  $\text{CH}_2\text{O}$  emitter. The model coefficients were determined by fitting the corrected  $\text{CH}_2\text{O}$  concentration data from carpet/cushion measurement Series I and II to the model. Substituting into Equation 3 for air exchange rates  $N_1$  and  $N_2$  under initial and final conditions, respectively,

$$[\text{CH}_2\text{O}]_1^{-1} = [(1.8 \cdot N_1 / 0.29) + 7.5] \quad (4)$$

$$[\text{CH}_2\text{O}]_2^{-1} = [(1.8 \cdot N_2 / 0.29) + 7.5] \quad (5)$$

Combining expressions, the  $N_2$  required to achieve an X-fold reduction in the  $\text{CH}_2\text{O}$  concentration measured at  $N_1$  is then,

$$N_2 = N_1 \cdot X + 1.22 \cdot X - 1.22 \quad (6)$$

The  $N_1 \cdot X$  term of Equation 5 represents simple ventilation theory that would predict a reduction in  $\text{CH}_2\text{O}$  concentration proportionate to the increase in air exchange rate. The second and third terms of Equation 5 result from the negative  $\text{CH}_2\text{O}$  concentration dependence of the  $\text{CH}_2\text{O}$  emission rate from pressed-wood products that has been modeled according to principles of molecular diffusion (see Equation 2). As the  $\text{CH}_2\text{O}$  concentration decreases with increased air exchange rates, the  $\text{CH}_2\text{O}$  emission rates from the solid  $\text{CH}_2\text{O}$  emission sources increase, reducing the impact of the enhanced ventilation. It is instructive to note that the second and third terms of Equation 6 are dependent on X (the reduction factor in  $\text{CH}_2\text{O}$  concentration) and/or  $\text{CH}_2\text{O}$  emission constants, but

independent of  $N_1$ . This is illustrated in Table 2 for X values of 2 and 4 with both low and high values of  $N_1$ , the initial air exchange rate. For large values of  $N_1$ ,  $N_2$  is approximately equal to  $N_1 \cdot X$ . This indicates that the decrease in  $\text{CH}_2\text{O}$  concentration with increased ventilation is insensitive to the  $\text{CH}_2\text{O}$  concentration dependence of the  $\text{CH}_2\text{O}$  emission rate from the pressed-wood products at the low  $\text{CH}_2\text{O}$  levels that are achieved with high air exchange rates. In contrast, at  $N_1$  values consistent with those frequently found in the research house (eg.,  $0.2 \text{ h}^{-1}$ ), the X dependent and constant terms of Equation 6 predominate. As a result, 8 and 24 fold increases in air exchange rate are required for twofold and fourfold reductions in  $\text{CH}_2\text{O}$  concentrations, respectively, with  $N_1$  equal to  $0.2 \text{ h}^{-1}$ .

The small reductions in  $\text{CH}_2\text{O}$  concentrations measured inside the Karns house with large increases in ventilation are exemplified in Figure 2. Approximate 1.2 and 1.6 fold reductions in  $\text{CH}_2\text{O}$  concentration were obtained for 2.5 and 5 fold increases in air exchange rate (from about 0.2 to 0.5, and  $1.0 \text{ h}^{-1}$ ). In contrast, the polyethylene sheet and vinyl linoleum barriers achieved approximate 1.8 and 2.1 fold reductions in  $\text{CH}_2\text{O}$  concentrations at an air exchange rate of about  $0.2 \text{ h}^{-1}$ . According to Equation 6, approximate six and eight fold increases in air exchange rate would be required for comparable improvements in  $\text{CH}_2\text{O}$  concentrations.

Table 2. Modeled increase in air exchange rate [ $N (\text{h}^{-1})$ ] from  $N_1$  to  $N_2$  required to achieve X fold reductions in  $\text{CH}_2\text{O}$  concentrations from  $[\text{CH}_2\text{O}]_1$  to  $[\text{CH}_2\text{O}]_2$ .

X	$N_1$	$N_2$	$N_2/N_1$
A. Low Air Exchange Rate Simulation			
1	---	---	1
2	0.2	1.6	8.1
2	0.5	2.2	4.4
2	1.0	3.2	3.2
4	0.2	4.5	22.3
4	0.5	5.7	11.3
4	1.0	7.7	7.7
B. High Air Exchange Rate Simulation			
2	2.0	5.2	2.6
2	5.0	11.2	2.2
4	2.0	10.4	5.8
4	5.0	23.7	4.7

## Conclusions

The effectiveness of (1) nylon carpet and neoprene sponge rubber cushion, (2) vinyl linoleum, and (3) 6 mil polyethylene sheet as  $\text{CH}_2\text{O}$  permeation barriers over particleboard underlayment has been studied using whole-house retrofits of floor coverings in an unoccupied research house. Although the carpet/cushion barrier had a minimal impact on the  $\text{CH}_2\text{O}$  emissions from the underlayment, the polyethylene and vinyl linoleum barriers reduced  $\text{CH}_2\text{O}$  emission rates measured specifically from the underlayment by 5 to 30 and 10 to 20 fold, respectively. However, the vinyl linoleum and polyethylene barriers resulted in only 2 to 2.5 fold reductions in indoor  $\text{CH}_2\text{O}$  concentrations. The comparatively small decrease in indoor levels is ascribed to secondary  $\text{CH}_2\text{O}$  emission sources inside the house whose emission rates are enhanced at the reduced  $\text{CH}_2\text{O}$  concentrations.

Approximate 1.6 fold reductions in  $\text{CH}_2\text{O}$  concentrations were achieved with a fivefold increase in air exchange rate from about 0.2 to  $1.0 \text{ h}^{-1}$ . Simple steady-state models incorporating single compartment, single emitter assumptions for the research house predict that sixfold and eightfold increases in air exchange rate would be needed for 2 to 2.5 fold reductions in  $\text{CH}_2\text{O}$  levels comparable to those achieved with the polyethylene and linoleum barriers.

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## References

- [1] J. G. Calvert, Chairman "Formaldehyde and other aldehydes", National Research Council, National Academy Press, Washington D.C., 1981.
- [2] T. G. Matthews, T. J. Reed, B. J. Tromberg, et al., "Formaldehyde emissions from combustion sources and solid formaldehyde resin containing products: potential impact on indoor formaldehyde concentrations and possible corrective measures", Proceedings of an Engineering Foundation Conference on Management of Atmospheres in Tightly Enclosed Spaces, p 23 (1984).
- [3] B. Meyer and K. Hermanns, "Formaldehyde release from pressed-wood products," in Formaldehyde, Analytical Chemistry, and Toxicology, American Chemical Society, Washington, DC, p. 101-16.
- [4] T. G. Matthews, T. J. Reed, B. J. Tromberg, et al., "Modeling and testing of formaldehyde emission characteristics of pressed-wood products", Report XVIII to the US Consumer Product Safety Commission, 1985.
- [5] American Society for Heating, Refrigeration and Air Conditioning Engineers, Standard 62-1981.

- [6] T. G. Matthews, K. W. Fung, B. J. Tromberg and A.R. Hawthorne, "Indoor air quality in the karns research houses: baseline measurements and impact of environmental parameters on formaldehyde concentrations", ORNL/TM-9433, 1985.
- [7] T. G. Matthews, T. J. Reed, K. W. Fung et al., "Draft environmental chamber protocol I: measurements of formaldehyde emissions from pressed-wood products", Appendix to Report XVII to the US Consumer Product Safety Commission, 1984.
- [8] A. R. Hawthorne, R. B. Gammage and C. S. Dudney, "An indoor air quality study of forty east tennessee homes", ORNL-5965, National Technical Information Service, Springfield, VA 1984.
- [9] T. G. Matthews, "Evaluation of a modified CEA Instruments inc. model 555 analyzer for the monitoring of formaldehyde vapor in domestic environments", Am. Ind. Hyg. Assoc. J 43(8): 547 (1982).
- [10] T. G. Matthews, A. R. Hawthorne, C. R. Daffron et al., "Formaldehyde surface emission monitor", Anal. Chem. 56: 448 (1984).
- [11] T. G. Matthews, T. J. Reed, C. R. Daffron et al., "Environmental dependence of formaldehyde emissions from pressed-wood products: experimental studies and modeling", Proceeding of the 18th International Washington State University Particleboard/Composite Materials Symposium, 1984, p. 41-70.
- [12] J. J. Hoetjer, "Introduction to a theoretical model for the splitting of formaldehyde from composition board", Report to Methanol Chemie Nederland, Delzijl (1978).

Appendix: Normalization factors for time-dependent decline in temperature, RH corrected CH<sub>2</sub>O concentration data for the Karns house

The CH<sub>2</sub>O concentration inside a single compartment with a single CH<sub>2</sub>O emission source, uniform mixing, constant temperature and RH at time zero (ie., [CH<sub>2</sub>O]<sub>I</sub>), and time final (ie., [CH<sub>2</sub>O]<sub>F</sub>), representing carpet/cushion measurement Series I and II, respectively, may be modeled as (11,12)

$$[\text{CH}_2\text{O}]^{-1} (\text{mg}/\text{m}^3) = A_I (\text{m}^2\text{h}/\text{mg}) \cdot N/L (\text{m}/\text{h}) + B_I (\text{m}^3/\text{mg}), \quad (\text{A1})$$

and

$$[\text{CH}_2\text{O}]_{F/N/L}^{-1} (\text{mg}/\text{m}^3) = A_F (\text{m}^2\text{h}/\text{mg}) \cdot N/L (\text{m}/\text{h}) + B_F (\text{m}^3/\text{mg}). \quad (\text{A2})$$

Assuming a linear decrease in CH<sub>2</sub>O concentrations with elapsed time, the modeled CH<sub>2</sub>O concentration at time Z (ie., [CH<sub>2</sub>O]<sub>Z</sub>) is

$$[\text{CH}_2\text{O}]_{Z/N/L} = [\text{CH}_2\text{O}]_{I/N/L} - (\text{Time } Z / \text{Total Time}) \cdot ([\text{CH}_2\text{O}]_{I/N/L} - [\text{CH}_2\text{O}]_{F/N/L}) \quad (\text{A3})$$

To normalize temperature and RH corrected CH<sub>2</sub>O concentration data for all barrier conditions (ie., [CH<sub>2</sub>O]<sub>B</sub>) to the carpet/cushion Series I measurements, the final expression is

$$[\text{CH}_2\text{O}]_{B \text{ Norm}} = ([\text{CH}_2\text{O}]_{I/N/L} / [\text{CH}_2\text{O}]_{Z/N/L}) \cdot [\text{CH}_2\text{O}]_B \quad (\text{A4})$$